

Mutual prodrug of cephazolin and benzylamin: 3-[(1-benzyl-1*H*-indazol-3-yl)oxy]-*N,N*-dimethylpropan-1-aminium 3-[(5-methyl-1,3,4-thiadiazol-2-yl)-sulfanyl]methyl]-8-oxo-7-[(1*H*-tetrazol-1-yl)acetamido]-5-thia-1-azabicyclo-[4.2.0]octane-2-carboxylate (benzydaminium cephazolinate)

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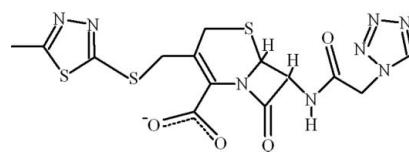
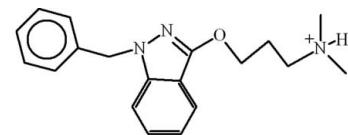
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Key indicators: single-crystal X-ray study; $T = 296\text{ K}$; mean $\sigma(\text{C}-\text{C}) = 0.005\text{ \AA}$; R factor = 0.050; wR factor = 0.095; data-to-parameter ratio = 14.2.

In the crystal of the title molecular salt, $\text{C}_{19}\text{H}_{24}\text{N}_3\text{O}^+\cdot\text{C}_{14}\text{H}_{13}\text{N}_8\text{O}_4\text{S}_3^-$, the cations and anions are linked by $\text{N}-\text{H}\cdots\text{O}$ hydrogen bonds. Short intramolecular $\text{C}-\text{H}\cdots\text{O}$ contacts occur within the anion and intermolecular $\text{C}-\text{H}\cdots\text{O}$ and $\text{C}-\text{H}\cdots\pi$ bonds help to establish the packing.

Related literature

Cephazolin, is a first generation cephalosporin antibiotic and benzylamin hydrochloride is a locally acting non-steroidal anti-inflammatory drug with local anaesthetic and analgesic properties. The title compound was prepared as a mutual prodrug for the treatment of infections and inflammatory conditions. For medicinal background to cephazolin, see: Turnbull (1995). For ring-puckering analysis, see: Cremer & Pople (1975). For graph-set theory, see: Bernstein *et al.* (1995).



Experimental

Crystal data



$M_r = 763.92$

Monoclinic, $C2$

$a = 44.409(3)\text{ \AA}$

$b = 7.1777(5)\text{ \AA}$

$c = 11.2683(8)\text{ \AA}$

$\beta = 90.587(8)^\circ$

$V = 3591.6(4)\text{ \AA}^3$

$Z = 4$

Mo $K\alpha$ radiation

$\mu = 0.27\text{ mm}^{-1}$

$T = 296\text{ K}$

$0.24 \times 0.14 \times 0.12\text{ mm}$

Data collection

Bruker Kappa APEXII CCD diffractometer

Absorption correction: multi-scan (*SADABS*; Bruker, 2005)

$T_{\min} = 0.957$, $T_{\max} = 0.968$

34321 measured reflections

6750 independent reflections

4092 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.088$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.050$

$wR(F^2) = 0.095$

$S = 1.00$

6750 reflections

475 parameters

1 restraint

H atoms treated by a mixture of independent and constrained refinement

$\Delta\rho_{\max} = 0.36\text{ e \AA}^{-3}$

$\Delta\rho_{\min} = -0.23\text{ e \AA}^{-3}$

Absolute structure: Flack (1983), 2918 Friedel Pairs

Flack parameter: $-0.07(7)$

Table 1
Hydrogen-bond geometry (\AA , $^\circ$).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{N}3-\text{H}3\text{N}\cdots\text{O}3^{\text{i}}$	0.96 (4)	1.70 (4)	2.660 (4)	174 (3)
$\text{N}7-\text{H}7\text{N}\cdots\text{O}2^{\text{ii}}$	0.86	2.02	2.868 (3)	167
$\text{C}18-\text{H}18\text{C}\cdots\text{O}4^{\text{iii}}$	0.96	2.52	3.303 (5)	139
$\text{C}24-\text{H}24\text{A}\cdots\text{N}4$	0.97	2.41	2.961 (5)	116
$\text{C}24-\text{H}24\text{B}\cdots\text{O}3$	0.97	2.38	2.882 (4)	112
$\text{C}32-\text{H}32\text{A}\cdots\text{O}2^{\text{iv}}$	0.97	2.28	3.218 (4)	164
$\text{C}32-\text{H}32\text{B}\cdots\text{O}2^{\text{ii}}$	0.97	2.50	3.318 (4)	142
$\text{C}19-\text{H}19\text{B}\cdots\text{CgA}$	0.96	2.91	3.869 (6)	174
$\text{C}20-\text{H}20\text{B}\cdots\text{CgD}$	0.97	2.72	3.642 (4)	158
$\text{C}29-\text{H}29\cdots\text{CgA}^{\text{ii}}$	0.98	2.99	3.861 (4)	149

Symmetry codes: (i) $x, y - 1, z + 1$; (ii) $x, y - 1, z$; (iii) $x, y, z + 1$; (iv) $-x, y - 1, -z + 1$. CgA and CgD are the centroids of the C1–C6 and N8–N11/C33 rings, respectively.

Data collection: *APEX2* (Bruker, 2007); cell refinement: *SAINT* (Bruker, 2007); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997) and *PLATON* (Spek, 2009); software used to prepare material for publication: *WinGX* (Farrugia, 1999) and *PLATON*.

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HB5219).

References

- Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N.-L. (1995). *Angew. Chem. Int. Ed. Engl.* **34**, 1555–1573.
- Bruker (2005). *SADABS*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Bruker (2007). *APEX2* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Cremer, D. & Pople, J. A. (1975). *J. Am. Chem. Soc.* **97**, 1354–1358.
- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
- Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.
- Flack, H. D. (1983). *Acta Cryst. A* **39**, 876–881.
- Sheldrick, G. M. (2008). *Acta Cryst. A* **64**, 112–122.
- Spek, A. L. (2009). *Acta Cryst. D* **65**, 148–155.
- Turnbull, R. S. (1995). *J. Can. Dent. Assoc.* **61**, 127–134.

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Acta Cryst. (2009). E65, o3132-o3133 [doi:10.1107/S1600536809047941]

Mutual prodrug of cephazolin and benzydamin: 3-[(1-benzyl-1*H*-indazol-3-yl)oxy]-*N,N*-dimethylpropan-1-aminium 3-{|[(5-methyl-1,3,4-thiadiazol-2-yl)sulfanyl]methyl}-8-oxo-7-[(1*H*-tetrazol-1-yl)acetamido]-5-thia-1-azabicyclo[4.2.0]octane-2-carboxylate (benzydaminium cephazolinate)

A. Asghar, M. S. Iqbal and M. N. Tahir

Comment

Cephazolin, is a first generation cephalosporin antibiotic and benzydamin hydrochloride is a locally acting nonsteroidal anti-inflammatory drug with local anaesthetic and analgesic properties providing both rapid and extended pain relief as well as a significant anti-inflammatory treatment for the painful inflammatory conditions of mouth and throat (Turnbull, 1995). The title compound (I, Fig. 1) was prepared as a mutual prodrug for the treatment of infections and inflammatory conditions.

No crystal structure has been found related to cation. In the cation the indazol moiety A (C1–C6/N2/N1/C14) and the benzene ring B (C8–C13) are planar with r.m.s. deviations of 0.0258 and 0.0031 Å respectively, from the respective mean square planes. The dihedral angle between A/B is 71.06 (12)°. In this cation, there exist an intramolecular H-bonding of C—H···O type (Table 1, Fig. 1) forming S(5) ring motif (Bernstein *et al.*, 1995) in the envelop form.

In the anion, two five membered heterocyclic rings C (C25/S3/C26/N5/N4) and D (N8—N9/C33) are planar with r. m. s. deviations of 0.0025 and 0.0014 Å respectively, from the respective mean square planes and the dihedral angle between C/D is 38.58 (14)°. The heterocyclic six membered E (C20/C21/C22/N6/C28/S1) is in the twisted form, with the maximum puckering amplitude $Q_T = 0.623$ (4) Å (Cremer & Pople, 1975) and the four membered ring F (N6/C28–C30) is not planar also. In anion, there exist two S(5) and a S(6) ring motifs (Table 1, Fig. 1). The molecules are stabilized in the form of polymeric chains forming R_2^1 (6), R_4^2 (8), R_2^2 (10) and other ring motifs (Fig. 2 & Fig. 3).

Experimental

Cefazolin sodium (0.238 g; 0.1 mol) and benzydamine hydrochloride (0.173 g; 0.01 mol) were dissolved separately in distilled water (50 ml). Ten milliliter of each solution were mixed and left for two days. The colourless needles of (I) thus obtained were filtered out by suction, washed with distilled water and dried under vacuum.

Refinement

The coordinates of H3N were refined. The H-atoms were positioned geometrically (O—H = 0.82 Å, N—H = 0.86 Å, C—H = 0.93–0.97 Å) and refined as riding with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{carrier})$ or $1.5U_{\text{eq}}(\text{methyl C})$.

supplementary materials

Figures

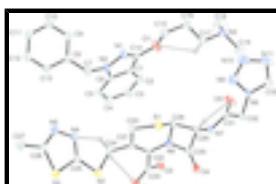


Fig. 1. View of (I) with displacement ellipsoids drawn at the 30% probability level.

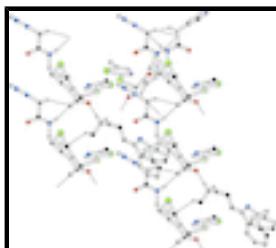


Fig. 2. The partial packing of (I), which shows that molecules form polymeric chains and ring motifs.

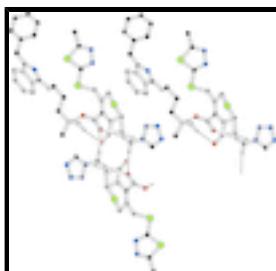


Fig. 3. The partial packing of (I), which shows that molecules form polymeric networks with various ring motifs.

3-[(1-Benzyl-1*H*-indazol-3-yl)oxy]-*N,N*-dimethylpropan-1-aminium 3-[[5-methyl-1,3,4-thiadiazol-2-yl)sulfanyl]methyl]-8-oxo-7-[*(1H*-tetrazol-1-yl)acetamido]-5-thia-1-azabicyclo[4.2.0]octane-2-carboxylate

Crystal data

$C_{19}H_{24}N_3O^+$	$C_{14}H_{13}N_8O_4S_3^-$	$F(000) = 1600$
$M_r = 763.92$		$D_x = 1.413 \text{ Mg m}^{-3}$
Monoclinic, $C2$		Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$
Hall symbol: C 2y		Cell parameters from 3824 reflections
$a = 44.409 (3) \text{ \AA}$		$\theta = 2.6\text{--}26.0^\circ$
$b = 7.1777 (5) \text{ \AA}$		$\mu = 0.27 \text{ mm}^{-1}$
$c = 11.2683 (8) \text{ \AA}$		$T = 296 \text{ K}$
$\beta = 90.587 (8)^\circ$		Cut needle, colourless
$V = 3591.6 (4) \text{ \AA}^3$		$0.24 \times 0.14 \times 0.12 \text{ mm}$
$Z = 4$		

Data collection

Bruker Kappa APEXII CCD diffractometer	6750 independent reflections
Radiation source: fine-focus sealed tube graphite	4092 reflections with $I > 2\sigma(I)$
Detector resolution: 7.60 pixels mm^{-1}	$R_{\text{int}} = 0.088$

ω scans $h = -54 \rightarrow 54$
 Absorption correction: multi-scan $k = -8 \rightarrow 8$
(SADABS; Bruker, 2005)
 $T_{\min} = 0.957, T_{\max} = 0.968$ $l = -13 \rightarrow 13$
 34321 measured reflections

Refinement

Refinement on F^2 Secondary atom site location: difference Fourier map
 Least-squares matrix: full Hydrogen site location: inferred from neighbouring sites
 $R[F^2 > 2\sigma(F^2)] = 0.050$ H atoms treated by a mixture of independent and constrained refinement
 $wR(F^2) = 0.095$ $w = 1/[\sigma^2(F_o^2) + (0.0342P)^2]$
 $S = 1.00$ where $P = (F_o^2 + 2F_c^2)/3$
 6750 reflections $(\Delta/\sigma)_{\max} < 0.001$
 475 parameters $\Delta\rho_{\max} = 0.36 \text{ e \AA}^{-3}$
 1 restraint $\Delta\rho_{\min} = -0.23 \text{ e \AA}^{-3}$
 Primary atom site location: structure-invariant direct Absolute structure: Flack (1983), 2918 Friedal Pairs
 methods Flack parameter: $-0.07 (7)$

Special details

Geometry. Bond distances, angles etc. have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

Refinement. Refinement of F^2 against ALL reflections. The weighted R -factor wR and goodness of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating R -factors(gt) etc. and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.14103 (6)	0.2863 (4)	0.9240 (3)	0.0552 (11)
N1	0.18241 (7)	0.2542 (4)	0.7971 (3)	0.0420 (12)
N2	0.19416 (7)	0.3698 (5)	0.7111 (3)	0.0433 (12)
N3	0.06627 (7)	0.0749 (5)	1.1012 (3)	0.0372 (11)
C1	0.15436 (8)	0.5181 (5)	0.7788 (3)	0.0340 (12)
C2	0.13236 (9)	0.6570 (6)	0.7813 (4)	0.0510 (17)
C3	0.13442 (11)	0.8000 (7)	0.7038 (4)	0.0640 (17)
C4	0.15844 (11)	0.8115 (6)	0.6251 (4)	0.0610 (19)
C5	0.18016 (10)	0.6769 (6)	0.6189 (3)	0.0490 (17)
C6	0.17762 (8)	0.5263 (5)	0.6969 (3)	0.0370 (12)
C7	0.21920 (9)	0.3080 (6)	0.6398 (3)	0.0463 (17)
C8	0.24923 (8)	0.3866 (5)	0.6760 (3)	0.0353 (12)

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C9	0.25878 (9)	0.3855 (6)	0.7933 (4)	0.0466 (17)
C10	0.28704 (10)	0.4478 (6)	0.8244 (4)	0.0520 (17)
C11	0.30587 (9)	0.5147 (6)	0.7391 (4)	0.0484 (17)
C12	0.29672 (9)	0.5182 (6)	0.6231 (4)	0.0492 (17)
C13	0.26851 (9)	0.4546 (6)	0.5911 (3)	0.0450 (17)
C14	0.15916 (9)	0.3469 (5)	0.8366 (3)	0.0393 (16)
C15	0.14483 (9)	0.1004 (6)	0.9660 (4)	0.0546 (17)
C16	0.12086 (10)	0.0662 (8)	1.0548 (4)	0.0699 (19)
C17	0.09024 (9)	0.0869 (7)	1.0079 (4)	0.0636 (19)
C18	0.06411 (12)	0.2420 (7)	1.1765 (4)	0.089 (3)
C19	0.03700 (9)	0.0436 (8)	1.0420 (4)	0.0718 (19)
S1	0.10575 (2)	0.19304 (14)	0.49823 (9)	0.0428 (3)
S2	0.14745 (2)	0.56694 (16)	0.18123 (9)	0.0463 (4)
S3	0.20738 (2)	0.44836 (15)	0.08659 (9)	0.0456 (4)
O2	0.05024 (6)	0.8249 (3)	0.3777 (2)	0.0384 (9)
O3	0.08047 (6)	0.7715 (4)	0.2253 (2)	0.0547 (11)
O4	0.02066 (6)	0.4136 (4)	0.3861 (2)	0.0405 (9)
O5	0.02552 (7)	0.0728 (4)	0.7592 (2)	0.0587 (13)
N4	0.19862 (7)	0.4956 (4)	0.3076 (3)	0.0406 (12)
N5	0.22865 (7)	0.4435 (4)	0.2966 (3)	0.0427 (12)
N6	0.06812 (6)	0.4815 (4)	0.4717 (2)	0.0279 (10)
N7	0.03965 (6)	0.0819 (4)	0.5676 (2)	0.0315 (10)
N8	0.02949 (7)	-0.3125 (4)	0.7545 (3)	0.0325 (11)
N9	0.05801 (8)	-0.3400 (5)	0.7855 (3)	0.0603 (16)
N10	0.05818 (10)	-0.4303 (7)	0.8833 (4)	0.0751 (17)
N11	0.02978 (11)	-0.4641 (5)	0.9184 (3)	0.0663 (16)
C20	0.13044 (8)	0.3791 (5)	0.4529 (3)	0.0440 (14)
C21	0.11584 (8)	0.5428 (5)	0.3900 (3)	0.0317 (12)
C22	0.08645 (8)	0.5831 (5)	0.3948 (3)	0.0293 (12)
C23	0.07099 (9)	0.7410 (5)	0.3273 (3)	0.0330 (12)
C24	0.13782 (8)	0.6622 (5)	0.3251 (3)	0.0396 (14)
C25	0.18490 (8)	0.5045 (5)	0.2061 (3)	0.0376 (12)
C26	0.23635 (9)	0.4134 (5)	0.1884 (3)	0.0407 (14)
C27	0.26722 (9)	0.3571 (6)	0.1520 (4)	0.0570 (17)
C28	0.07883 (8)	0.3496 (5)	0.5621 (3)	0.0347 (12)
C29	0.04609 (8)	0.2771 (5)	0.5704 (3)	0.0323 (12)
C30	0.04024 (8)	0.3968 (5)	0.4591 (3)	0.0289 (12)
C31	0.02943 (8)	-0.0044 (5)	0.6644 (3)	0.0340 (12)
C32	0.02179 (8)	-0.2078 (5)	0.6480 (3)	0.0337 (12)
C33	0.01254 (10)	-0.3891 (5)	0.8368 (4)	0.0479 (17)
H2	0.11664	0.65124	0.83513	0.0612*
H3	0.11965	0.89190	0.70279	0.0762*
H3N	0.0701 (8)	-0.035 (5)	1.148 (3)	0.0446*
H4	0.15971	0.91434	0.57520	0.0732*
H5	0.19589	0.68514	0.56533	0.0588*
H7A	0.21521	0.34164	0.55778	0.0555*
H7B	0.22030	0.17314	0.64381	0.0555*
H9	0.24592	0.34200	0.85174	0.0560*
H10	0.29334	0.44458	0.90336	0.0622*

H11	0.32494	0.55774	0.76019	0.0578*
H12	0.30960	0.56374	0.56540	0.0593*
H13	0.26244	0.45759	0.51190	0.0541*
H15A	0.16458	0.08557	1.00236	0.0653*
H15B	0.14299	0.01281	0.90075	0.0653*
H16A	0.12321	-0.05902	1.08589	0.0837*
H16B	0.12364	0.15230	1.12035	0.0837*
H17A	0.08661	-0.00932	0.94905	0.0766*
H17B	0.08860	0.20648	0.96821	0.0766*
H18A	0.05950	0.34834	1.12796	0.1331*
H18B	0.08295	0.26181	1.21710	0.1331*
H18C	0.04846	0.22478	1.23366	0.1331*
H19A	0.03248	0.14682	0.99069	0.1076*
H19B	0.02159	0.03237	1.10070	0.1076*
H19C	0.03785	-0.06890	0.99605	0.1076*
H7N	0.04237	0.01975	0.50338	0.0378*
H20A	0.14087	0.42565	0.52288	0.0529*
H20B	0.14552	0.32694	0.40083	0.0529*
H24A	0.15602	0.67492	0.37270	0.0474*
H24B	0.12927	0.78543	0.31426	0.0474*
H27A	0.27708	0.29376	0.21663	0.0852*
H27B	0.27858	0.46598	0.13134	0.0852*
H27C	0.26590	0.27544	0.08474	0.0852*
H28	0.08607	0.41003	0.63503	0.0418*
H29	0.03622	0.33403	0.63884	0.0385*
H32A	0.00044	-0.22075	0.63099	0.0402*
H32B	0.03281	-0.25736	0.58121	0.0402*
H33	-0.00840	-0.38952	0.83662	0.0575*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.058 (2)	0.0430 (17)	0.065 (2)	0.0076 (15)	0.0274 (16)	0.0174 (15)
N1	0.036 (2)	0.044 (2)	0.046 (2)	0.0012 (17)	0.0080 (18)	0.0076 (17)
N2	0.031 (2)	0.052 (2)	0.047 (2)	0.0036 (18)	0.0123 (17)	0.0080 (18)
N3	0.039 (2)	0.043 (2)	0.0295 (19)	0.0030 (18)	0.0015 (16)	0.0090 (17)
C1	0.030 (2)	0.037 (2)	0.035 (2)	-0.0035 (19)	-0.0037 (19)	0.0030 (19)
C2	0.049 (3)	0.050 (3)	0.054 (3)	0.008 (2)	0.007 (2)	0.009 (2)
C3	0.070 (3)	0.064 (3)	0.058 (3)	0.021 (3)	0.005 (3)	0.014 (3)
C4	0.082 (4)	0.048 (3)	0.053 (3)	0.003 (3)	0.002 (3)	0.013 (2)
C5	0.054 (3)	0.054 (3)	0.039 (3)	-0.015 (3)	0.001 (2)	0.010 (2)
C6	0.034 (2)	0.040 (2)	0.037 (2)	-0.006 (2)	-0.004 (2)	0.002 (2)
C7	0.042 (3)	0.051 (3)	0.046 (3)	-0.002 (2)	0.006 (2)	-0.003 (2)
C8	0.034 (2)	0.036 (2)	0.036 (2)	0.0023 (19)	0.005 (2)	0.003 (2)
C9	0.049 (3)	0.049 (3)	0.042 (3)	-0.007 (2)	0.008 (2)	0.001 (2)
C10	0.059 (3)	0.056 (3)	0.041 (3)	-0.008 (3)	-0.005 (2)	-0.002 (2)
C11	0.040 (3)	0.042 (3)	0.063 (3)	0.007 (2)	-0.003 (2)	0.000 (2)
C12	0.041 (3)	0.052 (3)	0.055 (3)	0.004 (2)	0.014 (2)	0.009 (2)

supplementary materials

C13	0.042 (3)	0.051 (3)	0.042 (3)	0.004 (2)	0.006 (2)	0.004 (2)
C14	0.031 (2)	0.047 (3)	0.040 (3)	-0.006 (2)	0.006 (2)	0.009 (2)
C15	0.042 (3)	0.054 (3)	0.068 (3)	0.005 (2)	0.010 (2)	0.015 (3)
C16	0.054 (3)	0.071 (3)	0.085 (4)	0.004 (3)	0.023 (3)	0.042 (3)
C17	0.051 (3)	0.082 (4)	0.058 (3)	-0.010 (3)	0.006 (3)	0.024 (3)
C18	0.128 (5)	0.061 (4)	0.077 (4)	0.019 (3)	-0.010 (4)	-0.030 (3)
C19	0.052 (3)	0.101 (4)	0.062 (3)	-0.009 (3)	-0.015 (2)	0.006 (3)
S1	0.0374 (6)	0.0314 (5)	0.0596 (7)	0.0060 (5)	0.0004 (5)	0.0109 (5)
S2	0.0375 (6)	0.0565 (7)	0.0448 (7)	-0.0009 (6)	0.0007 (5)	0.0019 (6)
S3	0.0463 (7)	0.0527 (7)	0.0380 (6)	0.0001 (6)	0.0054 (5)	-0.0031 (5)
O2	0.0384 (16)	0.0316 (15)	0.0452 (17)	0.0089 (13)	0.0084 (14)	0.0012 (13)
O3	0.067 (2)	0.0582 (19)	0.0393 (17)	0.0184 (16)	0.0200 (16)	0.0246 (15)
O4	0.0324 (15)	0.0485 (17)	0.0406 (16)	-0.0028 (13)	-0.0041 (14)	0.0086 (14)
O5	0.106 (3)	0.0393 (17)	0.0313 (16)	-0.0094 (17)	0.0191 (16)	-0.0010 (15)
N4	0.037 (2)	0.046 (2)	0.039 (2)	0.0003 (16)	0.0038 (16)	0.0039 (16)
N5	0.038 (2)	0.047 (2)	0.043 (2)	0.0016 (18)	0.0012 (17)	0.0032 (18)
N6	0.0282 (18)	0.0253 (17)	0.0302 (17)	-0.0001 (14)	0.0004 (14)	0.0099 (14)
N7	0.0414 (19)	0.0239 (17)	0.0293 (18)	-0.0033 (15)	0.0101 (15)	0.0011 (15)
N8	0.038 (2)	0.0287 (16)	0.0308 (19)	-0.0024 (16)	0.0055 (16)	0.0022 (16)
N9	0.044 (2)	0.080 (3)	0.057 (3)	0.010 (2)	0.000 (2)	0.016 (2)
N10	0.074 (3)	0.090 (3)	0.061 (3)	0.028 (3)	-0.009 (2)	0.022 (3)
N11	0.096 (3)	0.059 (3)	0.044 (2)	0.001 (3)	0.003 (2)	0.020 (2)
C20	0.036 (2)	0.043 (2)	0.053 (3)	-0.003 (2)	0.000 (2)	0.011 (2)
C21	0.031 (2)	0.027 (2)	0.037 (2)	-0.0021 (18)	0.0007 (18)	0.0008 (18)
C22	0.030 (2)	0.028 (2)	0.030 (2)	-0.0067 (18)	0.0040 (17)	0.0024 (18)
C23	0.036 (2)	0.028 (2)	0.035 (2)	-0.0032 (18)	-0.005 (2)	0.0028 (19)
C24	0.035 (2)	0.034 (2)	0.050 (3)	-0.0034 (19)	0.0070 (19)	0.004 (2)
C25	0.040 (2)	0.035 (2)	0.038 (2)	-0.0023 (19)	0.007 (2)	0.0035 (19)
C26	0.037 (2)	0.041 (2)	0.044 (3)	-0.001 (2)	0.003 (2)	-0.003 (2)
C27	0.044 (3)	0.062 (3)	0.065 (3)	0.006 (2)	0.008 (2)	0.001 (2)
C28	0.039 (2)	0.035 (2)	0.030 (2)	0.0001 (19)	0.0014 (19)	0.0055 (18)
C29	0.039 (2)	0.029 (2)	0.029 (2)	0.0000 (19)	0.0081 (18)	0.0008 (18)
C30	0.029 (2)	0.029 (2)	0.029 (2)	0.0037 (19)	0.0095 (19)	-0.0001 (18)
C31	0.036 (2)	0.036 (2)	0.030 (2)	-0.0018 (19)	0.0035 (19)	0.0063 (19)
C32	0.036 (2)	0.033 (2)	0.032 (2)	-0.0091 (18)	0.0000 (18)	0.0061 (18)
C33	0.055 (3)	0.048 (3)	0.041 (3)	-0.007 (2)	0.009 (2)	0.008 (2)

Geometric parameters (\AA , $^\circ$)

S1—C20	1.805 (4)	C11—C12	1.365 (6)
S1—C28	1.797 (4)	C12—C13	1.378 (6)
S2—C24	1.815 (4)	C15—C16	1.489 (6)
S2—C25	1.742 (4)	C16—C17	1.461 (6)
S3—C26	1.733 (4)	C2—H2	0.9300
S3—C25	1.732 (4)	C3—H3	0.9300
O1—C15	1.425 (5)	C4—H4	0.9300
O1—C14	1.350 (5)	C5—H5	0.9300
O2—C23	1.243 (5)	C7—H7B	0.9700
O3—C23	1.247 (4)	C7—H7A	0.9700

O4—C30	1.197 (4)	C9—H9	0.9300
O5—C31	1.217 (4)	C10—H10	0.9300
N1—N2	1.382 (5)	C11—H11	0.9300
N1—C14	1.310 (5)	C12—H12	0.9300
N2—C7	1.448 (5)	C13—H13	0.9300
N2—C6	1.351 (5)	C15—H15B	0.9700
N3—C17	1.506 (5)	C15—H15A	0.9700
N3—C18	1.473 (6)	C16—H16A	0.9700
N3—C19	1.472 (5)	C16—H16B	0.9700
N3—H3N	0.96 (4)	C17—H17A	0.9700
N4—N5	1.392 (4)	C17—H17B	0.9700
N4—C25	1.292 (5)	C18—H18B	0.9600
N5—C26	1.288 (5)	C18—H18A	0.9600
N6—C28	1.466 (4)	C18—H18C	0.9600
N6—C30	1.385 (4)	C19—H19C	0.9600
N6—C22	1.400 (4)	C19—H19B	0.9600
N7—C31	1.338 (4)	C19—H19A	0.9600
N7—C29	1.430 (5)	C20—C21	1.515 (5)
N8—N9	1.325 (5)	C21—C24	1.496 (5)
N8—C33	1.320 (5)	C21—C22	1.339 (5)
N8—C32	1.454 (5)	C22—C23	1.524 (5)
N9—N10	1.279 (6)	C26—C27	1.491 (6)
N10—N11	1.348 (7)	C28—C29	1.548 (5)
N11—C33	1.307 (6)	C29—C30	1.540 (5)
N7—H7N	0.8600	C31—C32	1.510 (5)
C1—C14	1.406 (5)	C20—H20A	0.9700
C1—C2	1.396 (5)	C20—H20B	0.9700
C1—C6	1.393 (5)	C24—H24A	0.9700
C2—C3	1.351 (7)	C24—H24B	0.9700
C3—C4	1.397 (7)	C27—H27A	0.9600
C4—C5	1.368 (6)	C27—H27B	0.9600
C5—C6	1.399 (5)	C27—H27C	0.9600
C7—C8	1.501 (5)	C28—H28	0.9800
C8—C9	1.384 (6)	C29—H29	0.9800
C8—C13	1.380 (5)	C32—H32A	0.9700
C9—C10	1.374 (6)	C32—H32B	0.9700
C10—C11	1.368 (6)	C33—H33	0.9300
C20—S1—C28	93.40 (17)	C15—C16—H16B	109.00
C24—S2—C25	100.75 (17)	C17—C16—H16B	109.00
C25—S3—C26	87.07 (17)	H16A—C16—H16B	108.00
C14—O1—C15	118.3 (3)	C17—C16—H16A	109.00
N2—N1—C14	103.8 (3)	C16—C17—H17A	109.00
C6—N2—C7	127.5 (3)	C16—C17—H17B	109.00
N1—N2—C6	111.9 (3)	H17A—C17—H17B	108.00
N1—N2—C7	120.2 (3)	N3—C17—H17B	109.00
C17—N3—C18	114.0 (4)	N3—C17—H17A	109.00
C17—N3—C19	108.6 (3)	N3—C18—H18B	109.00
C18—N3—C19	108.8 (4)	N3—C18—H18A	110.00
C17—N3—H3N	108 (2)	H18A—C18—H18C	109.00

supplementary materials

C18—N3—H3N	111 (2)	N3—C18—H18C	109.00
C19—N3—H3N	106 (2)	H18A—C18—H18B	109.00
N5—N4—C25	112.2 (3)	H18B—C18—H18C	109.00
N4—N5—C26	113.2 (3)	N3—C19—H19B	109.00
C22—N6—C30	133.6 (3)	N3—C19—H19C	109.00
C22—N6—C28	125.4 (3)	H19A—C19—H19C	109.00
C28—N6—C30	94.1 (3)	H19B—C19—H19C	109.00
C29—N7—C31	120.3 (3)	H19A—C19—H19B	109.00
N9—N8—C32	120.7 (3)	N3—C19—H19A	109.00
C32—N8—C33	131.6 (3)	S1—C20—C21	116.6 (3)
N9—N8—C33	107.6 (3)	C22—C21—C24	122.5 (3)
N8—N9—N10	107.5 (3)	C20—C21—C22	124.2 (3)
N9—N10—N11	110.3 (4)	C20—C21—C24	113.3 (3)
N10—N11—C33	105.2 (4)	N6—C22—C23	115.8 (3)
C29—N7—H7N	120.00	N6—C22—C21	119.1 (3)
C31—N7—H7N	120.00	C21—C22—C23	125.1 (3)
C6—C1—C14	103.5 (3)	O2—C23—C22	117.7 (3)
C2—C1—C6	120.5 (3)	O2—C23—O3	126.5 (3)
C2—C1—C14	135.8 (4)	O3—C23—C22	115.8 (3)
C1—C2—C3	118.5 (4)	S2—C24—C21	112.5 (2)
C2—C3—C4	120.8 (4)	S2—C25—N4	126.6 (3)
C3—C4—C5	122.3 (4)	S2—C25—S3	119.5 (2)
C4—C5—C6	116.9 (4)	S3—C25—N4	114.0 (3)
N2—C6—C1	107.0 (3)	S3—C26—C27	122.4 (3)
C1—C6—C5	120.9 (3)	S3—C26—N5	113.6 (3)
N2—C6—C5	132.1 (3)	N5—C26—C27	124.0 (4)
N2—C7—C8	114.8 (3)	N6—C28—C29	87.8 (2)
C7—C8—C9	121.4 (3)	S1—C28—C29	116.3 (2)
C7—C8—C13	120.0 (3)	S1—C28—N6	109.8 (2)
C9—C8—C13	118.6 (3)	N7—C29—C28	121.0 (3)
C8—C9—C10	120.8 (4)	N7—C29—C30	119.8 (3)
C9—C10—C11	119.9 (4)	C28—C29—C30	85.1 (3)
C10—C11—C12	120.1 (4)	O4—C30—N6	132.0 (3)
C11—C12—C13	120.4 (4)	N6—C30—C29	91.0 (3)
C8—C13—C12	120.3 (3)	O4—C30—C29	137.0 (3)
N1—C14—C1	113.8 (3)	O5—C31—C32	120.9 (3)
O1—C14—N1	124.2 (3)	O5—C31—N7	123.9 (3)
O1—C14—C1	122.1 (3)	N7—C31—C32	115.2 (3)
O1—C15—C16	107.1 (4)	N8—C32—C31	110.4 (3)
C15—C16—C17	114.2 (4)	N8—C33—N11	109.4 (4)
N3—C17—C16	113.8 (4)	S1—C20—H20A	108.00
C1—C2—H2	121.00	S1—C20—H20B	108.00
C3—C2—H2	121.00	C21—C20—H20A	108.00
C4—C3—H3	120.00	C21—C20—H20B	108.00
C2—C3—H3	120.00	H20A—C20—H20B	107.00
C3—C4—H4	119.00	S2—C24—H24A	109.00
C5—C4—H4	119.00	S2—C24—H24B	109.00
C6—C5—H5	122.00	C21—C24—H24A	109.00
C4—C5—H5	122.00	C21—C24—H24B	109.00

N2—C7—H7A	109.00	H24A—C24—H24B	108.00
C8—C7—H7B	109.00	C26—C27—H27A	109.00
N2—C7—H7B	109.00	C26—C27—H27B	109.00
C8—C7—H7A	109.00	C26—C27—H27C	110.00
H7A—C7—H7B	108.00	H27A—C27—H27B	109.00
C10—C9—H9	120.00	H27A—C27—H27C	109.00
C8—C9—H9	120.00	H27B—C27—H27C	109.00
C9—C10—H10	120.00	S1—C28—H28	113.00
C11—C10—H10	120.00	N6—C28—H28	113.00
C12—C11—H11	120.00	C29—C28—H28	113.00
C10—C11—H11	120.00	N7—C29—H29	110.00
C13—C12—H12	120.00	C28—C29—H29	110.00
C11—C12—H12	120.00	C30—C29—H29	110.00
C8—C13—H13	120.00	N8—C32—H32A	110.00
C12—C13—H13	120.00	N8—C32—H32B	110.00
O1—C15—H15B	110.00	C31—C32—H32A	110.00
C16—C15—H15A	110.00	C31—C32—H32B	110.00
O1—C15—H15A	110.00	H32A—C32—H32B	108.00
H15A—C15—H15B	109.00	N8—C33—H33	125.00
C16—C15—H15B	110.00	N11—C33—H33	125.00
C15—C16—H16A	109.00		
C20—S1—C28—C29	155.3 (3)	N8—N9—N10—N11	0.1 (5)
C20—S1—C28—N6	57.8 (3)	N9—N10—N11—C33	-0.1 (5)
C28—S1—C20—C21	-47.1 (3)	N10—N11—C33—N8	0.2 (5)
C24—S2—C25—N4	12.6 (4)	C2—C1—C6—N2	-176.6 (3)
C24—S2—C25—S3	-167.0 (2)	C2—C1—C6—C5	2.6 (6)
C25—S2—C24—C21	-111.3 (3)	C14—C1—C2—C3	-175.0 (4)
C25—S3—C26—C27	-179.3 (3)	C14—C1—C6—C5	178.4 (3)
C26—S3—C25—N4	-0.1 (3)	C2—C1—C14—O1	-5.2 (7)
C25—S3—C26—N5	-0.4 (3)	C14—C1—C6—N2	-0.8 (4)
C26—S3—C25—S2	179.6 (2)	C6—C1—C2—C3	-0.9 (6)
C14—O1—C15—C16	-176.1 (3)	C6—C1—C14—N1	-0.5 (4)
C15—O1—C14—C1	171.1 (3)	C2—C1—C14—N1	174.4 (4)
C15—O1—C14—N1	-8.4 (5)	C6—C1—C14—O1	179.9 (3)
N2—N1—C14—C1	1.4 (4)	C1—C2—C3—C4	-1.7 (7)
C14—N1—N2—C7	-175.3 (3)	C2—C3—C4—C5	2.7 (7)
N2—N1—C14—O1	-179.0 (3)	C3—C4—C5—C6	-1.0 (6)
C14—N1—N2—C6	-1.9 (4)	C4—C5—C6—C1	-1.6 (6)
N1—N2—C6—C5	-177.3 (4)	C4—C5—C6—N2	177.3 (4)
C7—N2—C6—C5	-4.6 (7)	N2—C7—C8—C9	48.2 (5)
N1—N2—C7—C8	-102.2 (4)	N2—C7—C8—C13	-134.6 (4)
C6—N2—C7—C8	85.6 (5)	C7—C8—C13—C12	-176.7 (4)
C7—N2—C6—C1	174.5 (3)	C9—C8—C13—C12	0.6 (6)
N1—N2—C6—C1	1.7 (4)	C13—C8—C9—C10	-1.1 (6)
C19—N3—C17—C16	-164.0 (4)	C7—C8—C9—C10	176.1 (4)
C18—N3—C17—C16	74.5 (5)	C8—C9—C10—C11	1.1 (6)
C25—N4—N5—C26	-0.7 (4)	C9—C10—C11—C12	-0.5 (7)
N5—N4—C25—S3	0.4 (4)	C10—C11—C12—C13	0.0 (7)
N5—N4—C25—S2	-179.2 (3)	C11—C12—C13—C8	0.0 (6)

supplementary materials

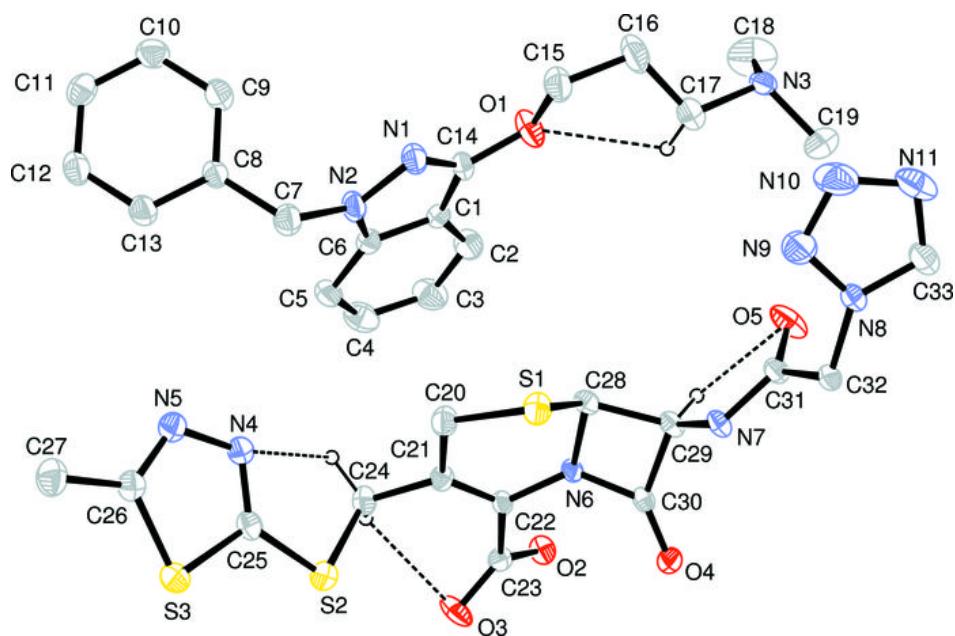
N4—N5—C26—S3	0.6 (4)	O1—C15—C16—C17	59.5 (5)
N4—N5—C26—C27	179.6 (3)	C15—C16—C17—N3	-173.0 (4)
C30—N6—C22—C21	-133.6 (4)	S1—C20—C21—C22	19.2 (5)
C30—N6—C22—C23	49.7 (5)	S1—C20—C21—C24	-163.3 (2)
C28—N6—C22—C23	-167.3 (3)	C20—C21—C22—N6	6.9 (5)
C30—N6—C28—C29	-11.3 (3)	C20—C21—C22—C23	-176.7 (3)
C22—N6—C28—S1	-48.2 (4)	C20—C21—C24—S2	81.3 (3)
C22—N6—C28—C29	-165.4 (3)	C22—C21—C24—S2	-101.2 (4)
C22—N6—C30—O4	-16.5 (7)	C24—C21—C22—N6	-170.3 (3)
C28—N6—C22—C21	9.4 (5)	C24—C21—C22—C23	6.0 (6)
C30—N6—C28—S1	106.0 (2)	N6—C22—C23—O2	32.8 (5)
C22—N6—C30—C29	161.9 (4)	C21—C22—C23—O3	37.0 (5)
C28—N6—C30—O4	-167.1 (4)	N6—C22—C23—O3	-146.5 (3)
C28—N6—C30—C29	11.3 (3)	C21—C22—C23—O2	-143.7 (4)
C29—N7—C31—O5	-1.4 (5)	S1—C28—C29—N7	21.1 (4)
C31—N7—C29—C30	-145.3 (3)	S1—C28—C29—C30	-100.9 (3)
C29—N7—C31—C32	176.6 (3)	N6—C28—C29—N7	132.1 (3)
C31—N7—C29—C28	111.5 (4)	N6—C28—C29—C30	10.1 (2)
N9—N8—C32—C31	-69.8 (4)	N7—C29—C30—O4	44.4 (6)
N9—N8—C33—N11	-0.1 (4)	N7—C29—C30—N6	-133.9 (3)
C32—N8—C33—N11	-177.9 (3)	C28—C29—C30—O4	167.5 (5)
C33—N8—N9—N10	0.0 (4)	C28—C29—C30—N6	-10.7 (3)
C32—N8—N9—N10	178.1 (4)	O5—C31—C32—N8	-38.1 (5)
C33—N8—C32—C31	107.8 (4)	N7—C31—C32—N8	143.9 (3)

Hydrogen-bond geometry (\AA , °)

$D—H\cdots A$	$D—H$	$H\cdots A$	$D\cdots A$	$D—H\cdots A$
N3—H3N···O3 ⁱ	0.96 (4)	1.70 (4)	2.660 (4)	174 (3)
N7—H7N···O2 ⁱⁱ	0.8600	2.0200	2.868 (3)	167.00
C18—H18C···O4 ⁱⁱⁱ	0.9600	2.5200	3.303 (5)	139.00
C24—H24A···N4	0.9700	2.4100	2.961 (5)	116.00
C24—H24B···O3	0.9700	2.3800	2.882 (4)	112.00
C32—H32A···O2 ^{iv}	0.9700	2.2800	3.218 (4)	164.00
C32—H32B···O2 ⁱⁱ	0.9700	2.5000	3.318 (4)	142.00
C19—H19B···CgA	0.96	2.91	3.869 (6)	174
C20—H20B···CgD	0.97	2.72	3.642 (4)	158
C29—H29···CgA ⁱⁱ	0.98	2.99	3.861 (4)	149

Symmetry codes: (i) $x, y-1, z+1$; (ii) $x, y-1, z$; (iii) $x, y, z+1$; (iv) $-x, y-1, -z+1$.

Fig. 1



supplementary materials

Fig. 2

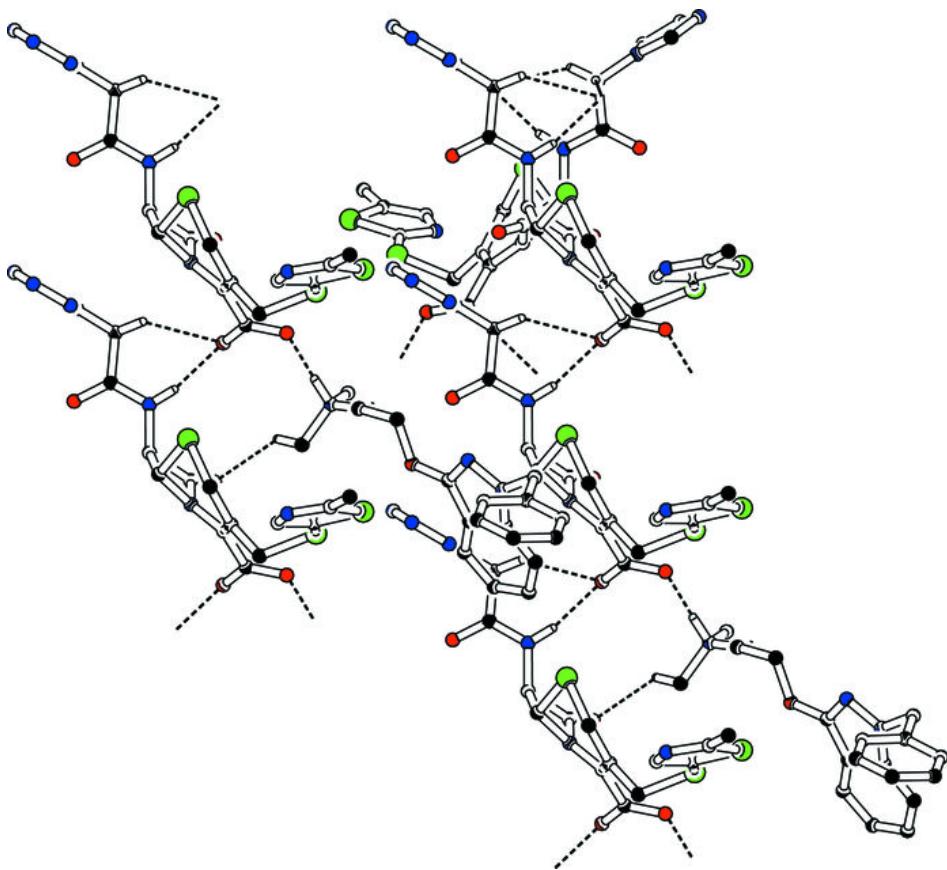


Fig. 3

